

The SL model: The SL (Solid/Liquid) model is the simplest of all material models and can be used for incompressible working substances, mostly pure solids and pure liquids. When a block of copper is heated or water is pumped by a centrifugal pump, the SL model can be used for property evaluation of the working substance.

Assumptions: All state equations for the SL model are derived from the following basic assumptions:

- (i) Specific volume v (and, hence, density $\rho = \frac{1}{v}$) is a constant, that is, the working substance is incompressible.
- (ii) Specific heat c_v is a constant (this is actually a corollary to the first assumption).

If we consider a metal block, it expands when heated; however, the same block, when idealized by the SL model, must be considered incompressible. Table A-1 and A-2 list material properties of several common solids and liquids. In the SL-state daemon, specific heats and specific volumes are populated as soon as a working substance is selected.

SL model equations: (Constant values of $\rho = 1/v$ and $c_p = c_v = c$ are read from tables)

$$\Delta u \equiv u_2 - u_1 = c(T_2 - T_1); \quad c_v = c_p = c; \quad (1)$$

$$\Delta h \equiv h_2 - h_1 = \Delta(u + pv) = \Delta u + \Delta(pv) = c(T_2 - T_1) + v(p_2 - p_1) \quad (2)$$

$$\Delta s = c_p \ln \frac{T_2}{T_1} \quad (3)$$

General state equations: (Applies to any substance)

$$m = \rho V; \quad \rho = \frac{1}{v}; \quad ke = \frac{V^2}{2000}; \quad pe = \frac{gz}{1000}; \quad e \equiv u + ke + pe; \quad j \equiv h + ke + pe; \quad h \equiv u + pv \quad (4)$$

$$E = me; \quad S = ms; \quad KE = m(ke); \quad PE = m(pe) \quad (5)$$

$$\dot{m} = \rho AV; \quad \dot{V} = AV; \quad \dot{E} = \dot{me}; \quad \dot{S} = \dot{ms} \quad (6)$$

$$Tds = du + pdv = dh - vdp; \quad c_v \equiv \left(\frac{\partial u}{\partial T} \right)_v; \quad c_p \equiv \left(\frac{\partial h}{\partial T} \right)_p \quad (7)$$

Reference: Chapter 1 introduces the concept of states and properties, Chapter 3 covers various material models and state evaluation, and Chapter 11 introduces advanced concepts on property evaluation. Read more about the SL model in Sec. 3.3.